Name:			
Date Started:	Date Completed:	Score:	



	Activity Sheet tals and Nonmetals		
A. Directions: Read and analyze each test item carefully. Circ	cle the letter of the correct answer.		
 Which of the following correctly describes pure substances? A. All pure substances can be decomposed into their simplest form. B. Pure substances are materials that have distinct characteristics. C. Pure substances like compounds have varying chemical compositions. D. Elements and compounds are pure substances that are hard to identify. 			
2. Which of the following is NOT a property of a metal? A. ductility B. insulator C. luster D. malleability			
 3. Which of the following statements does NOT describe a compound? A. It has a fixed chemical composition. B. It is made up of two or more elements. C. It contains an indefinite number of atoms. D. Its components can be separated by a chemical change. 			
4. Which does NOT belong to the group? A. aluminum B. carbon C. iron D. potassium			
 5. Which of the following is TRUE about nonmetals? A. They are generally solid. B. They have a high melting point. C. They have silvery or grayish luster. D. They are poor conductors of heat and electricity. 			
B. Directions. Classify each of them by writing \mathbf{M} for metal or $\mathbf{N}\mathbf{M}$ for nonmetal. Write your answer in the blank.			
1. carbon	6. iron		
2. bromine	7. phosphorus		
3. manganese	8. iodine		
4. radium	9. helium		

____10. platinum

____5. xenon